

hence were refined with unit occupancy factors of $1/3$. The fact that each pentane molecule has crystallographic $\bar{3}$ site symmetry gives rise to a badly disordered solvent molecule which could not be reliably modeled; the resulting large atomic esd's indicate that its molecular parameters are not physically meaningful.

All non-hydrogen atoms were refined anisotropically along with fixed contributions from the hydrogen atoms located at idealized positions with isotropic thermal parameters of $U = 0.08 \text{ \AA}^2$. Crystallographic computations were performed with SHELXTL-PLUS²⁰ software. The largest residual on the final difference map was only -0.54 e \AA^{-3} . Crystallographic data (Table 1), selected bond lengths and bond angles (Table 2), and atomic coordinates (Table 3) are presented.

It is apparent that the crystal packing is determined by the sterically demanding clusters, with the solvent molecules occupying the voids. Although a non-disordered crystal structure would have been highly desirable, crystals could not be obtained despite great effort. Of prime

(20) SHELXTL-PLUS, Siemens Analytical X-Ray Instruments, Inc.

importance, however, is that the structural determination unequivocally established the connectivities of all non-hydrogen atoms in **5**.

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Supplementary Material Available: Tables of structure determination data, anisotropic thermal parameters for non-hydrogen atoms, selected interatomic distances and angles, and idealized atomic parameters of hydrogen atoms for compound **5** (8 pages). Ordering information is given on any current masthead page.

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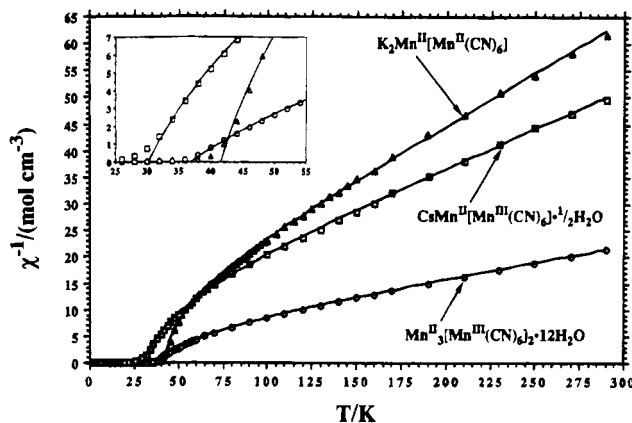
Additions and Corrections

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William R. Entley and Gregory S. Girolami*: New Three-Dimensional Ferrimagnetic Materials: $\text{K}_2\text{Mn}[\text{Mn}(\text{CN})_6]$, $\text{Mn}_3[\text{Mn}(\text{CN})_6]_2 \cdot 12\text{H}_2\text{O}$, and $\text{CsMn}[\text{Mn}(\text{CN})_6] \cdot \frac{1}{2}\text{H}_2\text{O}$.

Page 5165. In Table 1, the oxidation states for the metal centers in compound **2** were incorrect; the correct formulation is $\text{Mn}^{\text{II}}_3[\text{Mn}^{\text{III}}(\text{CN})_6]_2 \cdot 12\text{H}_2\text{O}$.

Page 5166. The degrees of hydration of compounds **2** and **3** in Figure 2 were incorrect. A revised version of Figure 2 is shown below.



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